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"A Bis(pyrazolyl) (bipyridyl) Platinum Complex"

by

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# A Bis(pyrazolyl) (bipyridyl) Platinum Complex

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**Abstract.** Bis(3,5-dimethylpyrazolium)4,4'-dimethyl-2,2'-bipyridyl platinum(II)·0.5 tetrahydrofuran solvate·H<sub>2</sub>O, PtC<sub>24</sub>H<sub>32</sub>N<sub>6</sub>O<sub>1.5</sub>, M<sub>r</sub> = 623.65; monoclinic, P2<sub>1</sub>/n; a = 8.625(2), b = 20.593(8), c = 14.451(4) Å, β = 90.32(2)°, V = 2566.7(14) Å<sup>3</sup>, Z = 4, D<sub>x</sub> = 1.61 g cm<sup>-3</sup>, MoKα, 0.71073Å, μ = 55.50 cm<sup>-1</sup>, F(000) = 1232, room temperature, R = 0.0387 for 2874 reflections with F<sub>o</sub><sup>2</sup> > 3σ(F<sub>o</sub><sup>2</sup>). The square-planar Pt complex has normal Pt-N(bipyridyl) bonds (2.009(8) Å) and slightly short Pt-N(pyrazolyl) bonds (1.983(7) Å). The ligand molecules have normal distances and angles; the planes of the pyrazolyl ligands are twisted by about 60° to the bipyridyl-Pt plane, with the closest contacts between the pyrazolyls being ~ 3.3 Å (C14···N5 and C19···N3).

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**Introduction.** In the course of our work on platinum(II) pyrazolyl bridged dimers, we have prepared a series of bis(pyrazolyl)bipyridyl platinum(II) monomers. These complexes have emissive states of MLCT or  $\pi$ - $\pi^*$  character, depending on the substituents on the pyrazole ring. Here we report the structure of bis(3,5-dimethylpyrazolium)-4,4'-dimethyl-2,2'-bipyridyl platinum(II), a derivative synthesized according to the literature method for the unmethylated analogue (Minghetti et al., 1979).

**Experimental.** Crystal a yellow needle,  $0.07 \times 0.07 \times 0.36$  mm; CAD-4 diffractometer,  $\omega$  scans; 25 reflections with  $14^\circ < 2\theta < 16^\circ$  used for unit cell; absorption correction based on psi scans of 6 reflections, relative transmissions from 0.882 to 1.000;  $(\sin\theta/\lambda)_{\max}$ , 0.59  $\text{\AA}^{-1}$ ;  $h$  from -10 to 10,  $k$  from -24 to 24,  $l$  from 0 to 17; three standard reflections (204,  $2\bar{5}\bar{2}$ ,  $2\bar{3}3$ ) showed no variations greater than predicted by counting statistics; 9834 reflections measured, 4501 independent; goodness of fit for merging 4368 multiples, 0.974;  $R_{\text{merge}} = 0.041$  for 3377 duplicates. All reflections used in solution and refinement of the structure; Pt atom located from Patterson map, remaining heavy atoms found by successive structure factor-Fourier calculations;  $F^2$  values used in least squares, with  $w = 1/\sigma^2(F_o^2)$ ; hydrogen atoms positioned by calculation ( $C-H = 0.95 \text{\AA}$ ) and not refined; coordinates and anisotropic displacement parameters of all atoms in the Pt molecule and the water O atom plus a scale factor refined;  $R$  (on  $F$ ) for 3933 reflections with  $F_o^2 > 0$ , 0.062;  $wR$  (on  $F^2$ ), 0.0078;  $S = 1.49$  for 271 parameters and 4501 reflections; weights taken as  $1/\sigma^2(F_o^2)$ ; variances ( $\sigma^2(F_o^2)$ ) derived from counting statistics plus an additional term,  $(0.014I)^2$ ; variances of the merged data by propagation of error plus another additional term,  $(0.014\bar{I})^2$ .  $(\Delta/\sigma)_{\max}$ , 0.01; final difference map has 1 peak  $2.2 \text{ e\AA}^{-3}$ ,  $1.8 \text{ \AA}$  from C18 and C19, next highest  $1.4 \text{ e\AA}^{-3}$  near the Pt atom; largest negative peak,  $-1.9 \text{ e\AA}^{-3}$ , near C23 of disordered THF. Atomic scattering factors and dispersion corrections from Cromer and Waber (1974) and Cromer (1974); computer programs were those of the CRYM Crystallographic Computing System (Duchamp, 1964) and ORTEP (Johnson, 1976). Final

refined parameters of the atoms are listed in Table 1. \* The tetrahydrofuran molecule is located near a center of symmetry and its parameters could not be refined; an idealized THF molecule was positioned based on difference maps.

We collected data for this compound with a crystal that had  $\beta = 90.14^\circ$ , solved and refined the structure, but large peaks in the difference map and distorted geometry in pyrazolyl ligand 2 caused us to conclude that our crystal was bad. The results reported here are based on data from a crystal that showed no sign of any twinning or deformity; still, the large positive peak in the difference map is near where the worst one was for the first crystal.

#### Discussion.

A drawing of the molecule including the numbering system is shown in Figure 1, and Table 2 gives distances and angles in the molecule. The packing is shown in Figure 2. The Pt-N distances to the bipyridyl N atoms are equal at 2.009(13) Å, and to the pyrazolyl N atoms at 1.983(11) Å. Distances and angles in the ligand atoms are normal, with C-CH<sub>3</sub> bonds being a bit short, especially in the pyrazolyl ligands (C18-C19, 1.449(14) Å is the shortest). The pyrazolyl ligands are twisted out of the Pt-bipyridyl plane by 64(3)° each. This orientation is comparable to other cis-bis nitrogen heterocycle platinum(II) systems (41.7° for cis-[Pt(N-methylimidazole)<sub>2</sub>Cl<sub>2</sub>] (Graves, Hodgson, van Kralingen & Reedijk, 1978) and 55.3° and 73.2° for cis-[Pt(pyrazole)<sub>2</sub>Cl<sub>2</sub>] (Cinelli *et al.*, 1989). There are short distances between N3 and C19 (3.32(1) Å) and N5 and C14 (3.33(1) Å). The hydrogen atoms on C14 and C19 show up as two sets of three H atoms each in the plane where they are expected, but none of them points toward the close nitrogen atom. The water

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\* Lists of assigned hydrogen parameters, anisotropic displacement parameters, complete distances and angles, and observed and calculated structure factors have been deposited with the British Library Document Supply Centre as Supplementary Publication No. SUP XXXXX (25 pp.). Copies may be obtained through The Executive Secretary, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, ENGLAND.

molecule is 2.82(1) Å from N4, indicating a hydrogen bond between them; there are also contacts between the water molecule and C5 and C8 in a different molecule (3.30(1) and 3.20(1) Å). The Pt–O(water) distance (4.492(7) Å) and all other intermolecular distances are at van der Waals' distances or greater.

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**Legends to Figures**

Figure 1. An ORTEP drawing of the molecule showing the numbering system. Heavy atoms are shown as 50% probability ellipsoids, hydrogen atoms as spheres of small, arbitrary size.

Figure 2. An ORTEP projection down the  $a$  axis, with 50% probability ellipsoids. The contents of one unit cell (not including hydrogen atoms) are shown, plus three additional THF molecules. Only one molecular orientation is shown at each THF site.

## **Supplementary Material for:**

## A Bis(pyrazolyl) (bipyridyl) Platinum Complex

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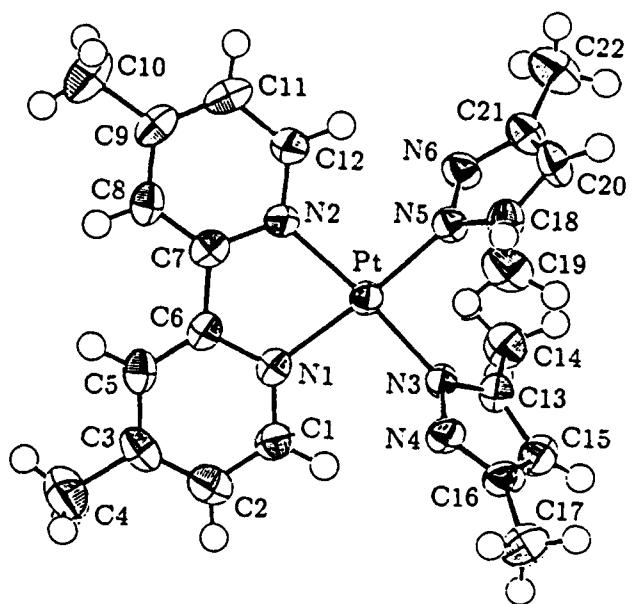


Figure 1. Schaefer, Connick, Miskowski  
& Gray

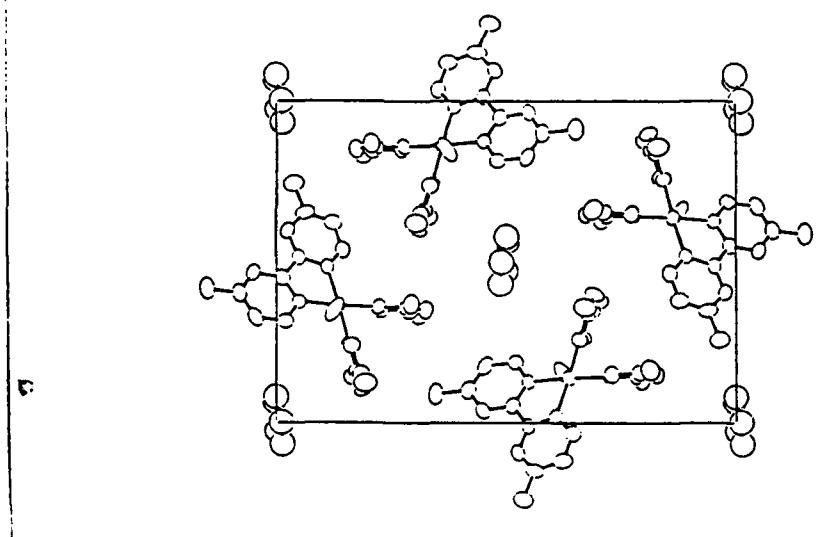


Figure 2. Schaefer, Connick, Mikowski  
& Gray

**Table 1. Final Refined Parameters for  
Bis(3,5-dimethylpyrazolium)(5,5'-dimethyl-2,2'-dipyridyl) Platinum(II).**

*x, y, z and  $U_{eq}^a \times 10^4$*

Atom	<i>x</i>	<i>y</i>	<i>z</i>	$U_{eq}$
Pt	2451(.4)	1366(.2)	3618(.3)	372(1)
N1	1470(8)	1091(3)	4821(5)	396(18)
C1	540(10)	1446(5)	5346(7)	491(24)
C2	-19(11)	1231(5)	6192(7)	579(29)
C3	380(11)	620(5)	6506(7)	519(29)
C4	-198(14)	363(6)	7413(8)	818(38)
C5	1326(11)	243(4)	5948(7)	486(25)
C6	1867(9)	480(4)	5124(7)	393(22)
C7	2904(9)	129(4)	4492(7)	412(24)
C8	3346(11)	-506(4)	4646(7)	495(25)
C9	4327(11)	-804(5)	4013(8)	537(27)
C10	4760(14)	-1510(5)	4131(9)	810(39)
C11	4858(11)	-452(5)	3263(8)	539(27)
C12	4342(11)	194(5)	3147(7)	488(25)
N2	3379(8)	481(3)	3747(5)	384(18)
N3	1428(8)	2224(3)	3584(5)	396(19)
C13	2057(10)	2817(4)	3578(7)	445(24)
C14	3747(11)	2936(5)	3683(8)	615(29)
C15	872(12)	3267(5)	3496(7)	567(28)
C16	-471(10)	2901(5)	3445(7)	501(26)

**Table 1. (Cont.)**

Atom	<i>x</i>	<i>y</i>	<i>z</i>	<i>U<sub>eq</sub></i>
C17	-2140(12)	3126(5)	3354(8)	735(34)
N4	-161(8)	2272(4)	3502(5)	454(21)
N5	3488(8)	1581(3)	2430(5)	413(19)
C18	2854(11)	1740(5)	1615(7)	487(26)
C19	1160(12)	1696(6)	1420(8)	709(34)
C20	4045(12)	1901(5)	1020(7)	589(30)
C21	5428(11)	1837(5)	1529(7)	498(26)
C22	7058(12)	1937(6)	1233(8)	805(38)
N6	5087(8)	1639(4)	2389(6)	489(21)
W1	7648(8)	1255(3)	3509(6)	901(23)

$$^a U_{eq} = \frac{1}{3} \sum_i \sum_j [U_{ij}(a_i^* a_j^*)(\vec{a}_i \cdot \vec{a}_j)]$$

**Table 2. Distances and Angles not Involving Hydrogen in  
Bis(3,5-dimethylpyrazolium)(5,5'-dimethyl-2,2'-dipyridyl) Platinum(II).**

		Distance(Å)	Distance(Å)	
Pt	-N1	2.018(7)	N3	-N4 1.379(10)
Pt	-N2	2.000(7)	C13	-C14 1.485(13)
Pt	-N3	1.975(7)	C13	-C15 1.384(13)
Pt	-N5	1.990(7)	C15	-C16 1.383(14)
N1	-C1	1.327(12)	C16	-C17 1.517(15)
N1	-C6	1.376(11)	C16	-N4 1.326(12)
C1	-C2	1.389(14)	N5	-C18 1.338(12)
C2	-C3	1.381(14)	N5	-N6 1.386(10)
C3	-C4	1.501(15)	C18	-C19 1.489(14)
C3	-C5	1.388(14)	C18	-C20 1.383(14)
C5	-C6	1.371(13)	C20	-C21 1.404(14)
C6	-C7	1.471(12)	C21	-C22 1.486(15)
C7	-C8	1.379(13)	C21	-N6 1.342(12)
C7	-N2	1.363(11)		
C8	-C9	1.391(14)		
C9	-C10	1.512(15)		
C9	-C11	1.384(14)		
C11	-C12	1.412(14)		
C12	-N2	1.341(12)		
N3	-C13	1.337(11)		

**Table 2. (Cont.)**

:

Angle(°)		Angle(°)	
N1 -Pt -N2	80.4(3)	C10 -C9 -C8	120.1(9)
N1 -Pt -N3	94.8(3)	C11 -C9 -C8	119.3(9)
N1 -Pt -N4	84.4(2)	C11 -C9 -C10	120.7(9)
N2 -Pt -N3	175.2(3)	C12 -C11 -C9	118.8(9)
N2 -Pt -N4	153.3(2)	N2 -C12 -C11	122.2(8)
N3 -Pt -N4	23.9(2)	C12 -N2 -C7	117.9(7)
C6 -N1 -C1	118.2(7)	N4 -N3 -C13	109.7(7)
C2 -C1 -N1	122.8(9)	C14 -C13 -N3	123.2(8)
C3 -C2 -C1	119.6(9)	C15 -C13 -N3	108.2(8)
C4 -C3 -C2	121.7(9)	C15 -C13 -C14	128.6(9)
C5 -C3 -C2	117.7(9)	C16 -C15 -C13	105.0(9)
C5 -C3 -C4	120.7(9)	C17 -C16 -C15	129.2(9)
C6 -C5 -C3	120.7(9)	N4 -C16 -C15	111.2(8)
C5 -C6 -N1	121.1(8)	N4 -C16 -C17	119.6(8)
C7 -C6 -N1	113.8(7)	C16 -N4 -N3	105.9(7)
C7 -C6 -C5	125.2(8)	N6 -N5 -C18	110.0(7)
C8 -C7 -C6	122.3(8)	C19 -C18 -N5	123.2(8)
N2 -C7 -C6	114.6(7)	C20 -C18 -N5	107.7(8)
N2 -C7 -C8	123.2(8)	C20 -C18 -C19	128.9(9)
C9 -C8 -C7	118.7(9)	C21 -C20 -C18	106.5(9)

**Table 2. (Cont.)**

Angle( $^{\circ}$ )		
C22	-C21	-C20
129.7(9)		
N6	-C21	-C20
108.9(8)		
N6	-C21	-C22
121.4(9)		
C21	-N6	-N5
106.8(7)		

**Table S1. Non-Refined Parameters for  
Bis(3,5-dimethylpyrazolium)(5,5'-dimethyl-2,2'-dipyridyl) Platinum(II).**

*x, y, z and U<sub>eq</sub><sup>a</sup> × 10<sup>4</sup>*

Atom	<i>x</i>	<i>y</i>	<i>z</i>	<i>B</i>	
O	5040	-30	660	10.0	*
C23	4770	20	-270	10.0	*
C24	6150	130	-670	10.0	*
C25	7280	110	10	10.0	*
C26	6630	-10	800	10.0	*
H1	253	1864	5135	4.5	*
H2	-665	1504	6554	5.3	*
H4A	205	-64	7508	7.6	*
H4B	169	636	7902	7.6	*
H4C	-1287	352	7415	7.6	*
H5	1587	-183	6135	4.3	*
H8	3009	-741	5171	4.5	*
H10A	4296	-1672	4683	7.3	*
H10B	4402	-1752	3617	7.3	*
H10C	5856	-1549	4183	7.3	*
H11	5555	-644	2837	4.9	*
H12	4699	434	2627	4.5	*
H14A	4273	2526	3723	5.7	*
H14B	4120	3169	3169	5.7	*
H14C	3942	3173	4237	5.7	*
H15	969	3730	3494	5.0	*
H17A	-2794	2755	3330	6.5	*
H17B	-2405	3387	3869	6.5	*
H17C	-2259	3370	2800	6.5	*
H19A	634	1576	1960	6.5	*
H19B	802	2110	1210	6.5	*
H19C	988	1384	944	6.5	*
H20	3939	2030	394	5.2	*
H22A	7735	1860	1739	7.4	*
H22B	7292	1658	738	7.4	*
H22C	7178	2380	1036	7.4	*
H23A	4332	-372	-499	10.0	*
H23B	4083	370	-393	10.0	*
H24A	6146	545	-958	10.0	*
H24B	6348	-195	-1121	10.0	*
H25A	8006	-223	-128	10.0	*
H25B	7800	516	39	10.0	*
H26A	6882	323	1230	10.0	*
H26B	6982	-416	1032	10.0	*

**Table S2. Anisotropic Displacement Parameters for Bis(3,5-dimethylpyrazolium)(5,5'-dimethyl-2,2'-dipyridyl) Platinum(II).**

Atom	$U_{11}$	$U_{22}$	$U_{33}$	$U_{12}$	$U_{13}$	$U_{23}$
Pt	306(2)	378(2)	432(2)	8(2)	-2(1)	44(2)
N1	381(42)	351(41)	455(49)	-27(33)	-60(37)	-15(36)
C1	441(53)	453(59)	579(66)	19(48)	33(49)	-17(54)
C2	524(61)	736(82)	477(66)	2(55)	38(51)	-34(59)
C3	394(58)	772(78)	392(63)	-150(53)	-30(49)	60(57)
C4	1039(98)	862(89)	554(79)	-144(73)	115(71)	72(67)
C5	518(61)	408(56)	531(68)	-95(47)	-58(53)	92(50)
C6	326(48)	347(51)	506(63)	-36(40)	-74(45)	14(46)
C7	260(47)	450(56)	524(65)	-30(40)	-115(45)	-33(49)
C8	531(61)	396(56)	558(68)	-64(47)	-44(53)	88(50)
C9	518(62)	422(59)	670(77)	97(50)	-172(57)	-57(57)
C10	989(90)	457(73)	980(98)	131(60)	-262(77)	-36(63)
C11	449(60)	488(62)	680(77)	131(48)	-111(55)	-179(57)
C12	529(61)	509(61)	426(62)	68(49)	52(50)	-3(49)
N2	354(42)	377(43)	421(48)	19(34)	42(37)	7(37)
N3	310(40)	409(45)	470(49)	18(33)	43(36)	83(37)
C13	477(63)	413(54)	446(59)	1(44)	-8(52)	58(48)
C14	501(62)	552(67)	792(84)	-98(51)	1(60)	41(59)
C15	585(68)	450(61)	667(77)	41(53)	21(58)	45(54)
C16	393(56)	618(68)	493(66)	153(50)	-23(49)	18(53)
C17	587(74)	768(78)	851(93)	207(59)	-25(63)	83(67)
N4	260(40)	491(49)	610(56)	34(35)	-29(38)	26(41)
N5	316(40)	498(49)	426(48)	9(33)	-11(36)	37(37)
C18	467(63)	512(59)	481(70)	-30(47)	-107(52)	90(49)
C19	450(64)	1012(91)	666(82)	-18(60)	-29(59)	68(68)
C20	636(70)	732(75)	400(64)	-65(57)	20(56)	186(55)
C21	429(58)	623(66)	442(64)	-110(49)	26(50)	42(52)
C22	593(76)	1137(101)	687(87)	-100(66)	131(66)	25(74)
N6	339(44)	617(52)	511(55)	10(36)	34(39)	44(42)
W1	727(49)	766(52)	1207(67)	-157(44)	-256(47)	561(52)

$U_{i,j}$  values have been multiplied by  $10^4$

The form of the displacement factor is:

$$\exp -2\pi^2(U_{11}h^2a^*{}^2 + U_{22}k^2b^*{}^2 + U_{33}\ell^2c^*{}^2 + 2U_{12}hka^*b^* + 2U_{13}h\ell a^*c^* + 2U_{23}k\ell b^*c^*)$$

**Table S3. Complete Distances and Angles for Bis(3,5-dimethylpyrazolium)(5,5'-dimethyl-2,2'-dipyridyl) Platinum(II).**

	Distance(Å)		Distance(Å)
Pt - N1	2.018(7)	C16 - C17	1.517(15)
Pt - N2	2.000(7)	C16 - N4	1.326(12)
Pt - N3	1.975(7)	C17 - H17A	0.951
Pt - N5	1.990(7)	C17 - H17B	0.946
N1 - C1	1.327(12)	C17 - H17C	0.950
N1 - C6	1.376(11)	N5 - C18	1.338(12)
C1 - C2	1.389(14)	N5 - N6	1.386(10)
C1 - H1	0.946	C18 - C19	1.489(14)
C2 - C3	1.381(14)	C18 - C20	1.383(14)
C2 - H2	0.951	C19 - H19A	0.938
C3 - C4	1.501(15)	C19 - H19B	0.956
C3 - C5	1.388(14)	C19 - H19C	0.951
C4 - H4A	0.955	C20 - C21	1.404(14)
C4 - H4B	0.956	C20 - H20	0.947
C4 - H4C	0.939	C21 - C22	1.486(15)
C5 - C6	1.371(13)	C21 - N6	1.342(12)
C5 - H5	0.945	C22 - H22A	0.947
C6 - C7	1.471(12)	C22 - H22B	0.940
C7 - C8	1.379(13)	C22 - H22C	0.962
C7 - N2	1.363(11)	O - C23	1.367
C8 - C9	1.391(14)	O - C26	1.386
C8 - H8	0.947	C23 - C24	1.345
C9 - C10	1.512(15)	C23 - H23A	0.950
C9 - C11	1.384(14)	C23 - H23B	0.950
C10 - H10A	0.954	C24 - C25	1.381
C10 - H10B	0.945	C24 - H24A	0.950
C10 - H10C	0.951	C24 - H24B	0.950
C11 - C12	1.412(14)	C25 - C26	1.298
C11 - H11	0.950	C25 - H25A	0.950
C12 - N2	1.341(12)	C25 - H25B	0.950
C12 - H12	0.952	C26 - H26A	0.950
N3 - C13	1.337(11)	C26 - H26B	0.950
N3 - N4	1.379(10)		
C13 - C14	1.485(13)		
C13 - C15	1.384(13)		
C14 - H14A	0.959		
C14 - H14B	0.942		
C14 - H14C	0.951		
C15 - C16	1.383(14)		
C15 - H15	0.958		

Table S3 (Cont.)

:

	Angle(°)		Angle(°)
H19B -C19 -H19A	110.0	H26B -C26 -H26A	109.5
H19C -C19 -H19A	110.4		
H19C -C19 -H19B	108.9		
C21 -C20 -C18	106.5(9)		
H20 -C20 -C18	126.4		
H20 -C20 -C21	127.1		
C22 -C21 -C20	129.7(9)		
N6 -C21 -C20	108.9(8)		
N6 -C21 -C22	121.4(9)		
H22A -C22 -C21	109.6		
H22B -C22 -C21	110.0		
H22C -C22 -C21	108.7		
H22B -C22 -H22A	110.6		
H22C -C22 -H22A	108.7		
H22C -C22 -H22B	109.3		
C21 -N6 -N5	106.8(7)		
C26 -O -C23	107.7		
C24 -C23 -O	106.8		
H23A -C23 -O	110.1		
H23B -C23 -O	110.1		
H23A -C23 -C24	110.1		
H23B -C23 -C24	110.1		
H23B -C23 -H23A	109.5		
C25 -C24 -C23	108.2		
H24A -C24 -C23	109.8		
H24B -C24 -C23	109.8		
H24A -C24 -C25	109.8		
H24B -C24 -C25	109.8		
H24B -C24 -H24A	109.5		
C26 -C25 -C24	109.0		
H25A -C25 -C24	109.6		
H25B -C25 -C24	109.6		
H25A -C25 -C26	109.6		
H25B -C25 -C26	109.6		
H25B -C25 -H25A	109.5		
C25 -C26 -O	108.0		
H26A -C26 -O	109.8		
H26B -C26 -O	109.8		
H26A -C26 -C25	109.8		
H26B -C26 -C25	109.8		

Table S3 (Cont.)

			Angle(°)		Angle(°)
N1	-Pt	-N2	80.4(3)	H10B -C10 -H10A	109.6
N1	-Pt	-N3	94.8(3)	H10C -C10 -H10A	109.0
N1	-Pt	-N4	84.4(2)	H10C -C10 -H10B	109.8
N2	-Pt	-N3	175.2(3)	C12 -C11 -C9	118.8(9)
N2	-Pt	-N4	153.3(2)	H11 -C11 -C9	120.2
N3	-Pt	-N4	23.9(2)	H11 -C11 -C12	121.1
C6	-N1	-C1	118.2(7)	N2 -C12 -C11	122.2(8)
C2	-C1	-N1	122.8(9)	H12 -C12 -C11	118.7
H1	-C1	-N1	118.4	H12 -C12 -N2	119.1
H1	-C1	-C2	118.8	C12 -N2 -C7	117.9(7)
C3	-C2	-C1	119.6(9)	N4 -N3 -C13	109.7(7)
H2	-C2	-C1	120.1	C14 -C13 -N3	123.2(8)
H2	-C2	-C3	120.3	C15 -C13 -N3	108.2(8)
C4	-C3	-C2	121.7(9)	C15 -C13 -C14	128.6(9)
C5	-C3	-C2	117.7(9)	H14A -C14 -C13	109.0
C5	-C3	-C4	120.7(9)	H14B -C14 -C13	110.0
H4A	-C4	-C3	109.1	H14C -C14 -C13	109.8
H4B	-C4	-C3	109.1	H14B -C14 -H14A	109.4
H4C	-C4	-C3	110.3	H14C -C14 -H14A	108.6
H4B	-C4	-H4A	108.5	H14C -C14 -H14B	110.1
H4C	-C4	-H4A	109.9	C16 -C15 -C13	105.0(9)
H4C	-C4	-H4B	109.8	H15 -C15 -C13	127.0
C6	-C5	-C3	120.7(9)	H15 -C15 -C16	128.0
H5	-C5	-C3	119.5	C17 -C16 -C15	129.2(9)
H5	-C5	-C6	119.8	N4 -C16 -C15	111.2(8)
C5	-C6	-N1	121.1(8)	N4 -C16 -C17	119.6(8)
C7	-C6	-N1	113.8(7)	H17A -C17 -C16	108.7
C7	-C6	-C5	125.2(8)	H17B -C17 -C16	109.8
C8	-C7	-C6	122.3(8)	H17C -C17 -C16	109.4
N2	-C7	-C6	114.6(7)	H17B -C17 -H17A	109.7
N2	-C7	-C8	123.2(8)	H17C -C17 -H17A	109.4
C9	-C8	-C7	118.7(9)	H17C -C17 -H17B	109.8
H8	-C8	-C7	121.8	C16 -N4 -N3	105.9(7)
H8	-C8	-C9	119.5	N6 -N5 -C18	110.0(7)
C10	-C9	-C8	120.1(9)	C19 -C18 -N5	123.2(8)
C11	-C9	-C8	119.3(9)	C20 -C18 -N5	107.7(8)
C11	-C9	-C10	120.7(9)	C20 -C18 -C19	128.9(9)
H10A	-C10	-C9	109.1	H19A -C19 -C18	109.7
H10B	-C10	-C9	109.9	H19B -C19 -C18	108.7
H10C	-C10	-C9	109.5	H19C -C19 -C18	109.1

Table S4. Observed and Calculated Structure Factors for  
Bis(3,5-dimethylpyrazolium)(5,5'-dimethyl-2,2'-dipyridyl) Platinum(II)

The columns contain, in order,  $k$ ,  $10F_{obs}$ ,  $10F_{calc}$  and  $10\left(\frac{F_{obs}^2 - F_{calc}^2}{\sigma F_{obs}^2}\right)$ . A minus sign preceding  $F_{obs}$  indicates that  $F_{obs}^2$  is negative.

Bis(pyrazolium)dipyridyl Platinum Complex.

Page 1

-10	k	1	-9	k	5	-8	k	3	10	278	137	23
1	530	449	25	0	426	492	-15	1	495	548	-21	5
2	403	397	1	1	379	392	-3	2	739	733	2	-8
3	639	597	15	2	-71	84	-5	3	713	714	0	k
4	177	88	10	3	289	306	-3	4	89	143	-6	8
5	169	210	-6	4	526	504	7	5	246	178	14	0
				5	227	261	-7	6	625	605	8	887
				6	257	278	-4	7	635	609	10	857
				7	-81	16	-3	8	362	332	9	-5
0	158	32	8	8	514	483	-10	9	187	175	2	1
1	578	566	4	9	389	424	-10	10	427	422	1	237
2	193	11	15					11	598	592	1	246
3	367	404	-10					12	391	385	1	164
4	222	154	10					13	-49	107	-6	-10
								14	225	247	-4	162
										387	408	-5
												0
												296
												287
												315
												322
												-1
												7
												208
												163
												-7
												468
												479
												-3
												294
												282
												-2
												387
												-5
												408
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												564
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												218
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## Bis(pyrazolium)dipyridyl Platinum Complex

Page	2		
4	1528	1539	- 5
5	347	347	0
6	745	769	- 12
7	126	5	10
8	972	965	3
9	280	299	- 6
10	748	749	0
11	- 109	21	- 7
12	422	440	- 6
13	- 109	35	- 7
14	687	679	3
15	- 160	77	- 17
16	223	275	- 12
17	- 102	70	- 7
18	583	590	- 2
19	- 88	30	- 4
	- 6	k	5
1	368	402	- 13
2	1251	1261	- 4
3	459	478	- 8
4	430	474	- 19
5	366	370	- 1
6	1014	1048	- 17
7	612	630	- 8
8	628	632	- 2
9	- 177	68	- 24
10	668	622	20
11	422	398	8
12	522	539	- 7
13	- 30	26	0
14	441	397	14
15	456	444	4
16	627	613	5
17	246	204	8
18	77	130	- 5
	- 6	k	8
0	294	370	- 19
1	690	684	3
2	149	172	- 4
3	712	739	- 13
4	416	429	- 5
5	1032	1056	- 12
6	231	258	- 7
7	344	394	- 18
8	218	241	- 5
9	987	1024	- 16
10	201	198	0
11	- 154	24	- 14
12	159	98	8
13	834	808	11
14	193	190	0
15	223	198	4
16	137	136	0
17	655	626	10
	- 6	k	7
1	332	366	- 11
2	403	401	0
3	813	860	- 23
4	- 106	115	- 15
5	510	525	- 6
6	450	456	- 2
7	886	896	- 4
8	283	294	- 3
9	149	173	- 4
10	33	201	- 21
11	713	714	- 9
12	117	178	- 9
13	146	91	6
14	147	151	0
15	776	723	20
16	244	282	- 8
17	222	286	- 13

### Bis(pyrazolium)dipyridyl Platinum Complex

Page 2

-6	k	8	5	536	506	9	10	349	391	-17	18	141	4	-9		
0	560	538	6	6	273	247	5	11	-112	20	19	394	439	-13		
1	152	249	-21	7	176	145	4	12	223	244	-5	-5	k	7		
2	236	211	5	8	330	354	-6	13	639	602	16					
3	328	345	-5	9	424	382	11	14	652	638	6					
4	903	929	-12	10	274	285	-2	15	-158	107	-22	0	838	813	8	
								16	253	172	17	1	907	891	8	
5	329	364	-11		-6	k	13	17	670	667	1	2	228	182	11	
6	343	367	-8		1	45	74	18	501	496	1	3	828	796	16	
7	-177	193	-40		2	619	605	4	19	217	229	-2	4	408	400	3
8	648	647	0		3	-248	79	-32	20	-65	148	-12	5	1129	1141	-6
9	343	368	-8		4	21	179	-13				6	-100	108	-14	
10	482	468	4		5	-40	0	0	-5	k	4	7	209	221	-3	
11	95	17	4		6	632	564	22	1	316	257	21	8	239	322	-26
12	333	306	7		7	151	49	8	2	1714	1699	7	10	182	140	-1
13	367	364	1						3	244	244	0	11	-8	112	7
14	531	538	-2		-5	k	1	4	505	504	0	12	242	307	-18	
15	103	35	4					5	113	127	-2	13	766	736	13	
16	157	208	-7		0	513	512	0	6	1396	1393	1	14	294	313	-5
					1	753	783	-18	7	241	112	29	15	-15	159	-13
	-6	k	9		2	402	376	13	8	800	801	0	16	-93	100	-9
1	-122	40	-9		3	528	531	-2	9	136	63	9	17	628	596	12
2	907	920	-6		4	973	994	-13	10	853	849	2	18	186	235	-9
3	120	226	-19		5	1079	1081	-1	11	288	283	1		-5	k	8
4	246	225	4		6	563	510	28	12	710	696	6				
5	-130	50	-10		7	327	326	0	13	144	169	-4				
6	778	804	-11		8	1011	973	21	14	506	468	15	1	79	167	-13
7	175	209	-6		9	1128	1150	-12	15	252	178	16	2	955	929	12
8	426	418	2		10	587	631	-23	16	737	729	3	3	616	574	18
9	128	138	-1		11	-47	5	-1	17	128	133	0	4	375	369	2
10	485	492	-2		12	336	394	-23	18	137	173	-5	5	-91	79	-9
11	-159	58	-14		13	950	935	7	19	43	18	0	6	936	969	-16
12	459	487	-9		14	543	496	20	20	482	504	-7	7	466	437	11
13	-163	105	-18		15	197	264	-17				8	405	427	-8	
14	516	436	25		16	240	225	3	-5	k	5	9	-141	88	-17	
15	-34	88	-3		17	660	662	0	0	1222	1217	1	10	497	510	-5
	-6	k	10		18	603	617	-5	1	631	588	22	11	281	305	-6
					19	368	366	0	2	-90	12	-6	12	535	544	-3
					20	197	53	16	3	264	291	-9	13	-113	12	-7
					21	309	320	-2	4	1408	1407	0	14	334	394	-18
0	408	456	-12		-5	k	2	5	601	595	0	15	395	371	-15	
1	525	502	8					6	768	766	0	16	522	523	0	
2	-139	57	-12					7	358	334	-13	17	-117	136	-15	
3	408	412	-1		1	1173	1228	-31	8	974	1000	-7	18	184	143	5
4	601	594	2		2	166	127	9	9	618	632	10		-5	k	9
5	686	707	-6		3	1253	1235	10	10	847	826	3	0	609	608	0
6	164	205	-7		4	287	262	9	11	130	103	17	1	-149	75	-17
7	130	155	-3		5	511	497	7	12	469	425	0	2	226	216	2
8	449	452	-1		6	76	27	4	13	519	518	0	3	102	140	-16
9	593	592	0		7	1665	1677	-6	14	565	584	1	4	1008	993	6
10	230	264	-7		8	399	365	15	15	154	142	1	5	151	203	-10
11	-111	96	-10		9	332	368	-15	16	249	257	-19	6	354	349	1
12	216	267	-10		10	-44	102	-9	17	155	253	-19	7	231	262	-7
13	433	458	-7		11	1086	1039	24	18	630	579	8	8	753	767	-6
	-6	k	11		12	162	141	4	19	225	173	9	9	-27	122	-8
					13	151	205	-12	20	-97	13	-4	10	410	479	-26
1	414	495	-28		14	151	126	4				11	134	13	9	
2	-59	63	-3		15	1001	1008	-2	-5	k	6	12	449	441	2	
3	773	747	10		16	216	130	16				13	143	240	-17	
4	-70	60	-4		17	521	509	4	1	890	891	0	14	573	531	15
5	309	317	-2		18	181	18	16	2	542	567	-12	15	-106	167	-18
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10	478	479	0	8	549	555	-3	9	263	293	-11	15	780	747	5		
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												20	116	154	-5	

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21	182	216	-6	18	266	253	3	17	780	765	6	0	451	441	3	
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8	593	560	12	18	723	703	9	19	459	316	44	20	-125	44	-8
9	318	279	9	20	-118	115	-15	20	735	734	0	21	457	452	1
10	333	395	-17	21	219	236	-3	21	198	158	6		3	k	8
11	-110	29	-5	22	492	504	-4	22	76	46	1				
12	352	392	-10	23	255	263	-1					1	646	587	30
13	33	215	-19					3	k	5		2	863	866	-1
				3	k	2		0	1597	1587	4	3	857	853	1
2	k	15		1	1638	1658	-11	1	671	694	-14	4	302	302	0
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2	182	221	-7	3	2143	2041	46	3	417	387	15	6	830	856	-14
3	441	430	3	4	595	585	6	4	1523	1535	-6	7	938	900	19
								5	598	561	17				

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9	15	101	-6	3	k	12	4	2047	1995	28	6	381	332	23
10	600	601	0	1	458	482	5	377	340	26	7	1528	1619	-49
11	622	619	1	2	366	402	-12	887	918	-26	8	289	249	15
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13	256	219	9	4	128	171	-6	1563	1583	-13	10	594	604	-5
14	261	300	-11	5	378	386	-2	2047	216	-27	11	1813	1815	0
15	543	599	-24	6	424	441	-6	1130	1127	2	12	539	517	10
16	526	514	4	7	724	752	-12	343	373	-19	13	151	216	-15
17	162	180	-3	8	160	180	-3	912	892	14	14	427	395	12
18	-83	108	-9	9	-59	100	-7	326	328	-1	15	897	910	-8
19	406	386	6	10	243	305	-15	1088	1091	-1	16	343	368	-8
20	510	474	11	11	515	513	0	138	69	12	17	357	324	10
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				13	83	171	-10	-129	11	-14	19	588	582	2
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2	70	142	-10					-102	115	-17	22	73	34	1
3	81	122	-5	3	k	13	21	51	53	0				
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7	140	177	-7	2	64	118	-5	1623	1600	12	2	435	463	-15
8	956	975	-9	3	208	142	11	1296	1362	-41	3	257	166	29
9	232	191	10	4	686	658	3	599	601	0	4	2245	2318	-31
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13	-53	90	-6	8	518	493	9	802	811	-5	8	1416	1413	1
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15	167	22	13	10	307	338	-8	903	930	-16	10	859	917	-33
16	311	263	12	11	-150	9	-10	915	955	-23	11	101	138	-6
17	-207	5	-22	12	229	295	-14	656	694	-21	12	691	728	-19
18	710	638	27	13	260	280	-4	208	160	11	13	622	618	1
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5	96	209	-21	5	-125	73	-10	290	248	9	21	187	185	0
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7	731	690	19	7	235	163	12							
8	529	525	1	8	331	375	-12	4	k	2	1	155	194	-10
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7	304	342	-12	12	2	-93	80	444	423	7		4	k	6
8	199	174	4	13	447	403	12	667	129	-11				
9	997	974	10	14	-241	44	-27	503	457	15				
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15	271	269	0	20	2453	2463	5	1114	1060	31	5	1470	1476	-3
16	72	98	-1	21	100	80	4	1114	1060	31	6	176	184	-2
17	527	455	22	22	245	225	11	428	386	22	7	96	215	-27

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9	1006	1019	-7	18	128	111	1	1	-126	105	-12	8	282	278	1	
10	418	411	2					2	-113	71	-8	9	67	62	0	
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13	693	700	-3		0	995	1010	5	207	207	0	12	162	152	2	
14	505	534	-12		1	503	507	6	144	220	-11	13	108	101	0	
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16	253	265	-3		3	377	415	8	482	484	0	15	892	872	9	
17	637	650	-5		4	881	878	9	217	144	10	16	162	154	1	
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					8	774	792	-8				20	153	197	-7	
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7	1194	1202	-3		16	284	240	9	5	k	0	4	672	649	13	
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8	706	736	-14	7	118	63	4	11	154	26	12	8	759	707	22	
9	728	723	2	8	526	580	-12	12	533	542	-3	9	209	192	3	
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5	506	484	14	8	553	515	16	5	249	190	13	11	78	26	2	
6	270	286	-4	9	567	524	18	6	499	498	0	12	270	368	-24	
				10	615	623	-3	7	170	43	14					

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7	k	10	7	335	302	9	1	647	610	13	1	474	441	10	
1	182	217	-6	8	172	131	6	2	145	18	9	2	342	374	-9
2	578	570	2	9	666	662	1	3	779	798	-7	3	700	718	-7
3	303	335	-8	10	367	391	-7	4	-86	3	-2	4	13	94	-4
4	-145	120	-16	11	140	85	6	5	198	257	-11	5	248	287	-9
5	-113	4	-6	12	280	243	8	6	141	35	8	6	202	209	-1
6	513	490	7	13	759	700	23	7	774	777	-1	7	608	590	6
7	272	288	-3	14	298	220	16	8	-35	48	-1	8	59	204	-18
8	354	321	8	15	67	156	-6	9	235	216	3	9	-51	14	-1
9	-223	66	-25	8	k	3		10	134	104	3	10	332	256	17
10	361	415	-15	11	705	724	-8	11	691	577	39	11	617	607	3
7	k	11	2	288	252	-9	8	k	8		9	k	3		
0	250	278	-4	3	824	813	5	0	467	477	-2	1	235	65	5
1	427	434	-1	4	-38	54	-2	1	554	563	-3	2	626	584	5
2	-160	15	-11	5	389	353	11	2	157	88	7	3	89	87	1
3	290	289	0	6	298	217	19	3	353	368	-4	4	426	43	1
4	170	218	-7	7	786	766	0	4	485	476	2	5	208	195	2
5	469	471	0	8	183	51	15	5	520	508	3	6	671	683	-4
6	95	23	3	9	-110	64	-8	6	290	285	1	7	176	48	13
7	37	157	-9	10	198	167	5	7	98	55	2	8	235	128	16
8	183	115	8	11	944	917	11	8	479	488	-2	9	191	78	13
12	7	k	12	12	285	299	-3	9	808	605	0	10	676	663	4
13	14	212	79	16	105	81	2	10	200	177	3		145	144	0
1	392	361	8	8	k	4					9	k	4		
2	212	223	-1								1	397	409	-3	
3	385	417	-8	0	833	813	6	1	80	122	-9	2	595	614	-7
4	-210	127	-26	1	368	395	-8	2	752	773	-8	3	448	391	17
8	k	0	2	181	114	-9	3	-113	114	-11	4	141	207	-10	
1	4	771	766	12	4	172	136	4	5	173	146	3			
0	1075	1095	-9	5	406	413	-2	6	410	474	-19	6	529	512	5
1	259	153	33	6	286	323	-10	7	-88	22	-3	7	369	399	-9
2	317	259	24	7	247	233	3	8	401	405	0	8	356	323	6
3	191	48	27	8	682	670	5				9	-58	98	-5	
4	1029	1020	6	9	482	439	14	8	k	10		10	323	335	-3
5	256	213	15	10	551	522	10				9	k	5		
6	430	429	0	11	-127	90	-12	0	631	552	18				
7	150	96	10	12	499	474	8	1	187	229	-6	0	764	768	0
8	866	852	9	13	482	375	32	2	87	59	1	1	-157	25	-11
9	283	224	20	14	539	489	16	3	79	215	-16	2	82	35	2
10	581	537	23					4	513	500	4	3	118	4	6
11	92	29	5	8	k	5		5	242	183	9	4	699	688	4
12	482	495	-6								5	165	59	10	
13	271	213	17	1	221	77	20	9	k	0	6	243	276	-11	
14	570	541	14	2	937	922	6				7	-189	1	-17	
15	-85	85	-9	3	286	250	8	1	251	173	22	8	599	569	10
8	k	1	4	219	215	0	2	861	864	-1	9	250	51	24	
1	300	211	22	5	110	49	4	3	215	192	6				
2	955	933	10	6	736	743	-3	4	221	242	-6	9	k	6	
3	518	549	-13	7	175	185	-1	5	158	123	7				
4	460	414	17	8	352	333	5	6	724	740	-9	1	374	356	4
5	182	113	11	9	134	47	7	7	142	147	-1	2	467	488	-6
6	829	820	4	10	655	631	9	8	328	329	0	3	419	417	0
7	447	401	16	11	126	150	-3	9	65	55	0	4	198	122	10
8	408	384	8	12	534	533	0	10	542	495	23	5	199	176	3
9	113	244	-24	13	257	55	25	11	247	236	3	6	446	398	13
10	452	457	-1	8	k	6		9	k	1		7	498	464	10
11	365	291	20	0	660	640	5	0	832	759	20				
12	646	621	10	1	550	561	-4	1	294	269	6	0	111	24	3
13	40	24	0	2	-139	66	-12	2	49	107	-4	1	534	473	18
14	317	292	6	3	362	381	-5	3	216	179	6	2	-91	40	-4
15	157	217	-9	4	467	535	-25	4	759	756	1	3	345	366	-5
8	k	2	5	525	554	-10	5	352	365	-3	4	53	58	0	
7	6	-112	139	6	98	122	-2	6	297	257	9	5	585	592	-2
0	686	704	-5	8	439	479	-13	7	239	152	15				
1	735	728	3	9	545	489	19	8	566	542	8	9	k	8	
2	165	51	13	10	86	215	-17	9	267	201	13	1	369	416	-12
3	434	420	5	11	136	51	7	10	511	474	12	2	505	379	24
4	372	370	0	12	314	290	5	11	236	18	23				
5	868	817	23					9	k	2					
6	178	190	-2	8	k	7						10	k	0	

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	0	691	697	-2	10	k	1	10	k	2	4	314	255	13
1	203	142	13		1	288	195	17			10	k	3	
2	131	81	6		2	645	651	-2	0	205	243	-4		
3	215	140	18		3	284	249	7	1	492	490	0	1	492
4	712	665	24		4	167	186	-2	2	135	26	7	2	249
5	279	191	23		5	255	151	17	3	313	278	7	3	512